

Intramolecular Cyclization of β -Alkynylpropanamides to γ -Alkylidene- γ -butyrolactams

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Abstract. A general method for the base-catalized intramolecular cyclization of β -alkynylpropanamides 1 to γ -alkylidene- γ -butyrolactames 2 (and 3) was established. Reactions of β -alkynylamides 1c-h, possessing alkyl groups at the terminal acetylenes, in the presence of a catalytic amount of LiN(TMS)₂ / AgOTf (= 2:1) in toluene gave exclusively (Z)-alkylidenelactames 2c-h in good yields. © 1998 Elsevier Science Ltd. All rights reserved.

The intramolecular addition of nitrogen nucleophiles to acetylenic triple bonds, giving rise to a variety of N-heterocycles, is of importance in organic synthesis ¹. In many cases, the synthesis of cyclic amines from alkynylamines (or amides) has been conducted using bases ² or transition metals ³. Many indoles have also been obtained in this manner ⁴. Little study has been directed to the synthesis of lactams by the intramolecular cyclization of alkynylamides: e.g., Bu₄NF- or LiAl(NHBn)₄-catalyzed cyclization of alkynylamides ⁵, cyclization of ω -phenylseleno-substituted alkynylamides by Bu⁴OK-18-crown-6 ⁶, or the synthesis of 1,3-oxazolidine-2-ones by base-catalyzed cyclization of alkynylcarbamates in the presence of Ag-, Cu-salt or Pd⁷. In the case of β -alkynylamides (e.g.1c-h) possessing alkyl groups at terminal acetylenes, no such cyclization has been reported. The intramolecular cyclization of ω -alkynylbenzamide by Et₃N in the presence of Ag₂CO₃ was previously shown to afford a mixture of lactam and iminolactone (ratio 1:1) but when conducted with LiN(TMS)₂ in THF the lactam compound was obtained predominantly ⁸. The authors thus sought to establish a general method for the base-catalyzed intramolecular cyclization of β -alkynylamides (R = H, Ar, alkyl) ⁹ and the results summarized in Table 1 are discussed in the following.

By the above method (i.e., LiN(TMS)₂ in THF at 66 °C for 18 h)⁸⁾, the cyclization of aryl-substituted alkynylamide **1a** was conducted to provide moderate yield of **2a** and **3a** (Run 1). When done in DMF, the yield increased but product isolation from DMF was tedius (Run 4). KN(TMS)₂ / 18-crown-6 in THF at room temperature gave the best results (Run 3). With **1b** having a bulky *N*-substituted group, LiN(TMS)₂ treatment alone in DMF at 60°C afforded **3b** predominantly as the thermodynamically stable product (Run 6). Under these conditions or using bases in the presence of phase-transfer-catalyst or absence, alkyl-substituted alkynylamide **1c** could make no progress (Run 7). With the catalytic LiN(TMS)₂ / AgOTf (= 2:1) system in toluene^{10, 11)}, the cyclization of **1c** proceeded more efficiently (Run 8). THF or DME instead of toluene as the solvent had no significant effect (Run 10). This catalytic system used with various alkyl-substituted alkynylamides (R¹= chloropropyl, dodecyl, H: **1g**, **1h**, **1i**, respectively) or alkynylamides having bulky *N*-substituted alkyl groups (**1d**, **1e**, **1f**) led to satisfactory yields and stereoselectivity (Runs 9, 11-15)^{10, 12)}.

Table 1. Intramolecular Cyclization of β -Alkynylpropanamides to γ -Alkylidene- γ -butyrolactams Under Basic Conditions.

$$R^{1}$$
 basic conditions R^{1} R^{2} R^{2} R^{2} R^{2} R^{2} R^{2} R^{2} R^{2} R^{2}

12-1				2a-1				3a-i	
Run	Alkynylamide 1a-i R ¹ , R ²	Base (eq)	Additive (eq)	Solvent	Temp.	Time (h)	Yield $(\%)^a$ and Isomer Ratio $(2:3)^b$		
1	1a: R ¹ =p-methoxyphenyl R ² =m-methoxybenzyl	LHMDS (1.0)		THF	66	18	42	(85:15)	
2	1a :	n-BuLi (1.0)		THF	0-66	10	8		
3	1a :	KHMDS (0.5)	18-crown-6 (0.4)	THF	r.t.	3	71	(64 : 36)	
4	1a :	LHMDS (1.0)		DMF	60-65	3	64	(55:45)	
5	1a :	LHMDS (0.3)	AgOTf (0.15)	THF	65-70	3	85	(96: 4)	
6	1b: R ¹ =p-methoxyphenyl R ² =CH(Ph)CH ₂ OTBS (R)	LHMDS (1.0)		DMF	60-65	3	67	(12:88)	
7	1c: R ¹ =CH ₃ R ² =m-methoxybenzyl	KHMDS (0.3)	18-crown-6 (0.3)	THF	r.t60	18	0		
8	1c:	LHMDS (0.3)	AgOTf (0.15)	toluene	65-70	3	89	(100:0)	
9	$\mathbf{1d} \colon \mathbf{R}^1 = \mathbf{CH}_3$ $\mathbf{R}^2 = \mathbf{CH}(\mathbf{Ph})\mathbf{CH}_3 (S)$	LHMDS (0.3)	AgOTf (0.15)	toluene	65-70	3	89	(100:0)	
10	1d:	LHMDS (0.3)	AgOTf (0.15)	THF	66	18	46	(100:0)	
11	1e: R^1 =CH ₃ R^2 =CH(Ph)CH ₂ OCH ₃ (R)	LHMDS (0.3)	AgOTf (0.15)	toluene	65-70	3	88	(100:0)	
12	1f: R^1 =CH ₃ R^2 =CH(Ph)CH ₂ O $-$ (R)	LHMDS (0.3)	AgOTf (0.15)	toluene	65-70	4	85	(100:0)	
13	$\mathbf{1g}: \mathbf{R}^1 = \mathbf{n} - \mathbf{C}_3 \mathbf{H}_6 \mathbf{C} \mathbf{I}$ $\mathbf{R}^2 = m - \mathbf{methoxybenzyl}$	LHMDS (0.3)	AgOTf (0.15)	toluene	65-70	3	84	(100:0)	
14	1h: $R^1 = n - C_{12}H_{25}$ $R^2 = m$ -methoxybenzyl	LHMDS (0.3)	AgOTf (0.15)	toluene	65-70	4	89	(100:0)	
15	1i: $R^1 = H$ $R^2 = m$ -methoxybenzyl	LHMDS (0.3)	AgOTf (0.15)	toluene	65-70	3	86		

^a Isolated yield after purification by NH-silicagel column.

^b Determined based on ^lH NMR spectra of the crude products.

Further, when 1a was applied to this system, 2a was obtained in 82% yield along with 3a in 3% yield (Run 5). However, this method with 5-hexynamide failed to bring about 6-membered ring formation. Compound 2 was found to have the Z-form structure by NOE¹³⁾, as was also supported by examination of the isomerization of 2 to thermodynamically stable product 3 (*E*-form) in CDCl₃ for several hours¹⁴⁾ or on standing at room temperature for 2-3 days.

The reaction mechanism for the present cyclization remains to be clarified. The LiN(TMS)₂ / AgOTf (= 1:1) system in toluene gave no product, thus suggesting the mechanism in scheme 1. Reaction of 1 with silver- and lithium-amides, prepared from a mixture of 1 eq AgOTf and 2 eq LiN(TMS)₂, produced bis-metallated complex 4, which underwent *trans*-aminometallation to vinylmetal-species 5, followed by the protonolysis of 5 to 2. However, the cyclization of 5-hexynamide to δ -valerolactam under the same conditions failed to occur, the reasons for which, at present, are not understood.

Scheme 1
$$\begin{array}{c} AgOTf + 2LiN(TMS)_2 \\ \hline \\ N-R^2 \end{array} \xrightarrow{AgN(TMS)_2 \cdot LiN(TMS)_2 + LiOTf} \begin{array}{c} (TMS)_2NAg \\ \hline \\ R^2-N \\ \hline \\ Li O \end{array} \xrightarrow{R^1} \begin{array}{c} [Ag] \\ \hline \\ R^2 \end{array} \xrightarrow{N} \begin{array}{c} 2 \\ \hline \\ R^2 \end{array}$$

The authors have thus established a new method for the efficient intramolecular cyclization of β -alkynylamides to γ -alkylidenelactams. It is particularly significant that alkyl-substituted alkynylamides by the catalytic LiN(TMS)₂ / AgOTf system could efficiently undergo intramolecular cyclization to produce alkylidenelactams. The alkylidenelactam obtained in the present study should prove useful for the synthesis of α -substituted pyrrolidine derivatives.

References and Notes

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- 9. **1a, b, i** were prepared in the present study in the same manner as described in the previous paper⁸⁾. **1d-h** were prepared as outlined below.

Synthesis of IV (R¹= Me) from I by a little different conditions was reported in a total yield of 35%; Carling, R.W.; Clark, J.S.; Holmes, A.B.; Sartor, D. J. Chem. Soc., Perkin Trans I 1992, 95-101.

- 10. Synthesis of **2e** (general procedure for **1c-h**). To a solution of alkynylamide **1e** (400 mg, 1.630mmol) and AgOTf (64 mg, 0.24 mmol) in abs. toluene (6 ml) stirred under argon atmosphere was added slowly at r.t. a solution of LiN(TMS)₂ in hexane (1.0 M, 0.49 ml, 0.49 mmol). After 0.5 h, the mixture consisting of a clear solution and black-brown paste, was stirred at 65-70°C for 3 h, whereupon, the system became a black suspension. The reaction mixture was quenched with ice-water and filtered through celite by suction. The filtrate was extracted with AcOEt according to a conventional work-up. The crude product was purified by chromatography on NH-silica gel (eluent: hexane / AcOEt = 20:1) to give the **2e** (353 mg) in 88 % yield as a colorless solid. mp 73-74°C (from *i*Pr₂O-hexane). IR (KBr, cm⁻¹) 2900, 1670, 1320. ¹H-NMR (300 MHz, CDCl3) δ ; 1.53 (d, J = 7.4 Hz, 3H), 2.48 (m, 2H), 2.66 (m, 2H), 3.42 (s, 3H), 4.00 (dd, J = 5.9, 9.7 Hz, 1H), 4.35 (dd, J = 8.2, 9.7 Hz, 1H) 4.52 (q, J = 7.4 Hz, 1H), 5.38 (dd, J = 5.9, 8.2 Hz, 1H) 7.32 (m, 5H). ¹³C-NMR (75 MHz, CDCl₃) 12.3, 27.9, 30.3, 58.5, 58.7, 72.8, 96.8, 126.6, 127.2, 128.4, 138.4, 139.6, 178.0; MS m/z 235 (M⁺); [α]²⁴_D +70.1° (c 1.02, toluene). *Anal.* Calcd for C₁₅H₁₉NO₂: C, 73.44; H, 7.81; N, 5.71. Found: C, 73.42; H, 7.81; N, 5.68.
- 11. Based on the catalysis of Cu⁺ and Ag⁺ with appropriate bases, the intramolecular addition of 2-propynyl carbamates to their own acetylenic triple bonds has been reported ^{7d. 7f)}.
- 12. Synthesis of **2i**. A mixture of LiN(TMS)₂ / AgOTf (2:1) in toluene was added *via* syringe at r.t. to a solution of **1i** in toluene to form a black-brown precipitate gradually, followed by the same work-up described for general procedure to afford **2i** in 86% yield.
- 13. NOESY (500 MHz NMR, pyridine- d_5) demonstrated **2d** to have the Z-form as evident by the NOE interaction between 4-H and vinyl-H, vinyl-CH₃ and benzyl-H, and vinyl-CH₃ and NCH(CH₃).
- 14. Isomer **3d** was determined to have the *E*-form by NOESY subsequent to the isomerization of **2d** in CDCl₃ at r.t. for several hours.